

This Page Is Inserted by IFW Operations
and is not a part of the Official Record

BEST AVAILABLE IMAGES

Defective images within this document are accurate representations of the original documents submitted by the applicant.

Defects in the images may include (but are not limited to):

- BLACK BORDERS
- TEXT CUT OFF AT TOP, BOTTOM OR SIDES
- FADED TEXT
- ILLEGIBLE TEXT
- SKEWED/SLANTED IMAGES
- COLORED PHOTOS
- BLACK OR VERY BLACK AND WHITE DARK PHOTOS
- GRAY SCALE DOCUMENTS

IMAGES ARE BEST AVAILABLE COPY.

**As rescanning documents *will not* correct images,
please do not report the images to the
Image Problem Mailbox.**



INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification ⁶ : D06M 11/79, 15/277, 15/657	A1	(11) International Publication Number: WO 97/00995 (43) International Publication Date: 9 January 1997 (09.01.97)
(21) International Application Number: PCT/US96/08302 (22) International Filing Date: 3 June 1996 (03.06.96) (30) Priority Data: 60/000,468 23 June 1995 (23.06.95) US 08/653,548 24 May 1996 (24.05.96) US (71) Applicant: MINNESOTA MINING AND MANUFACTURING COMPANY [US/US]; 3M Center, P.O. Box 33427, Saint Paul, MN 55133-3427 (US). (72) Inventors: ENGLE, Lori, P.; P.O. Box 33427, Saint Paul, MN 55133-3427 (US). HAMROCK, Steven, J.; P.O. Box 33427, Saint Paul, MN 55133-3427 (US). PELLERITE, Mark, J.; P.O. Box 33427, Saint Paul, MN 55133-3427 (US). (74) Agents: WEISS, Lucy, C. et al.; Minnesota Mining and Manufacturing Company, Office of Intellectual Property Counsel, P.O. Box 33427, Saint Paul, MN 55133-3427 (US).	(81) Designated States: AU, CA, JP, European patent (AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE). Published <i>With international search report. Before the expiration of the time limit for amending the claims and to be republished in the event of the receipt of amendments.</i>	
(54) Title: COMPOSITION AND PROCESS FOR IMPARTING DURABLE REPELLENCY TO SUBSTRATES (57) Abstract A fluorochemical treatment composition comprises (a) a sol comprising a colloidal dispersion of substantially discrete, substantially monodisperse in size, polymer-grafted, inorganic microparticles in liquid; and (b) at least one fluorochemical. The composition can be used to impart durable water and oil repellency, stain release, and dry soil resistance to a variety of substrates.		

FOR THE PURPOSES OF INFORMATION ONLY

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AM	Armenia	GB	United Kingdom	MW	Malawi
AT	Austria	GE	Georgia	MX	Mexico
AU	Australia	GN	Guinea	NE	Niger
BB	Barbados	GR	Greece	NL	Netherlands
BE	Belgium	HU	Hungary	NO	Norway
BF	Burkina Faso	IE	Ireland	NZ	New Zealand
BG	Bulgaria	IT	Italy	PL	Poland
BJ	Benin	JP	Japan	PT	Portugal
BR	Brazil	KE	Kenya	RO	Romania
BY	Belarus	KG	Kyrgyzstan	RU	Russian Federation
CA	Canada	KP	Democratic People's Republic of Korea	SD	Sudan
CF	Central African Republic	KR	Republic of Korea	SE	Sweden
CG	Congo	KZ	Kazakhstan	SG	Singapore
CH	Switzerland	LI	Liechtenstein	SI	Slovenia
CI	Côte d'Ivoire	LK	Sri Lanka	SK	Slovakia
CM	Cameroon	LR	Liberia	SN	Senegal
CN	China	LT	Lithuania	SZ	Swaziland
CS	Czechoslovakia	LU	Luxembourg	TD	Chad
CZ	Czech Republic	LV	Latvia	TG	Togo
DE	Germany	MC	Monaco	TJ	Tajikistan
DK	Denmark	MD	Republic of Moldova	TT	Trinidad and Tobago
EE	Estonia	MG	Madagascar	UA	Ukraine
ES	Spain	ML	Mali	UG	Uganda
FI	Finland	MN	Mongolia	US	United States of America
FR	France	MR	Mauritania	UZ	Uzbekistan
GA	Gabon			VN	Viet Nam

COMPOSITION AND PROCESS FOR IMPARTING
DURABLE REPELLENCY TO SUBSTRATES

5 Field of the Invention

 This invention relates to compositions and processes for imparting water and oil repellency and/or stain and soil resistance to substrates such as, e.g., carpets, textiles, paper, and leather.

10

Background of the Invention

 Fluorochemicals are widely used to impart water and oil repellency as well as dry soil resistance to substrates, particularly fibrous substrates such as
15 carpets. The fluorochemicals thereby enable the substrates to retain their original aesthetic appeal. But many fluorochemical treatments lack durability.

 For example, fluorochemicals which have high glass transition temperatures can flake from treated
20 substrates when subjected to abrasion during use. This results in unprotected areas of substrate which thereafter fail to resist staining and soiling. Even fluorochemical treatments which are less brittle and prone to flaking often cannot withstand cleaning
25 operations.

 Thus, there is a need in the art for fluorochemical treatments which can provide durable water and oil repellency and durable dry soil resistance. Such treatments should preferably also
30 possess durable stain release properties, and, in view of current environmental concerns, the treatments should be deliverable from water rather than from organic solvent.

Summary of the Invention

Briefly, in one aspect, this invention provides a fluorochemical treatment composition which can be used to impart durable water and oil repellency, durable stain release, and durable dry soil resistance to substrates, e.g., carpets, fabrics (both woven and non-woven), papers, wood, concrete, and leather. The composition comprises (a) a sol comprising a colloidal dispersion of substantially discrete, substantially monodisperse in size, polymer-grafted, inorganic microparticles in liquid; and (b) at least one fluorochemical. Microparticles having polymodal size distributions prepared from blends of monodisperse distributions can also be utilized in forming the sol. The sol preferably comprises an aqueous dispersion of polymer-grafted silica microparticles, the polymer comprising polymerized units of at least one free-radically polymerizable, ethylenically-monounsaturated monomer. Preferably, the fluorochemical comprises at least one fluoroaliphatic or fluoroalicyclic group and is attached to the microparticles. As used herein, "polymer-grafted microparticles" are microparticles having linear or branched, substantially uncrosslinked polymer chains covalently bonded at essentially only one end to the microparticle (through surface-attached coupling agent).

The composition of the invention can be used to impart surprisingly durable water and oil repellency, stain release, and dry soil resistance to a variety of substrates. And, since the composition can be water-based, it satisfies the need in the art for durable fluorochemical treatments which are environmentally friendly.

In other aspects, the invention also provides one-step and two-step treatment processes for imparting

5 durable water and oil repellency, stain release, and
dry soil resistance properties; a treatment composition
useful for imparting durable stain release and dry soil
resistance, the composition comprising a sol comprising
10 a colloidal dispersion of substantially discrete,
substantially monodisperse in size, polymer-grafted,
inorganic microparticles in liquid (preferably, water
or liquid comprising water; more preferably, water); a
treatment process for providing substrates with durable
15 stain release and dry soil resistance properties; and
treated articles comprising the treatment compositions
of the invention or the corresponding dry compositions.

Detailed Description of the Invention

15 Inorganic microparticles suitable for use in the
compositions of the invention are colloidal in size
(e.g., having an average particle diameter in the range
of from about 1 nanometer (1 millimicron) to about 200
nanometers. Colloidal silica is preferred, but other
20 colloidal metal oxides, e.g., colloidal titania,
colloidal alumina, colloidal zirconia, colloidal
vanadia, colloidal chromia, colloidal iron oxide,
colloidal antimony oxide, colloidal tin oxide, and
mixtures thereof, can also be utilized. The colloidal
25 microparticles can comprise essentially a single oxide
such as silica or can comprise a core of an oxide of
one type (or a core of a material other than a metal
oxide) on which is deposited an oxide of another type.
Generally, the microparticles can range in size
30 (average particle diameter) from about 1 nanometer to
about 200 nanometers, preferably from about 1 nanometer
to about 100 nanometers, more preferably from about 2
nanometers to about 75 nanometers.

35 It is also preferable that the colloidal
microparticles be relatively uniform in size (have a

substantially monodisperse size distribution or a polymodal distribution obtained by blending two or more substantially monodisperse distributions) and remain substantially non-aggregated (substantially discrete), as microparticle aggregation can result in precipitation, gellation, or a dramatic increase in sol viscosity and can reduce both adhesion to substrate and optical clarity. Thus, a particularly desirable class of microparticles for use in preparing the compositions of the invention includes sols of inorganic microparticles (e.g., colloidal dispersions of inorganic microparticles in liquid media), especially sols of amorphous silica. Unlike fumed silica, which comprises irregular aggregates of colloidal particles, such sols are substantially monodisperse in size and shape and thus enable the preparation of compositions exhibiting good optical clarity and smoothness and, surprisingly, good adhesion to substrates. Such sols can be prepared by a variety of techniques and in a variety of forms which include hydrosols (where water serves as the liquid medium), organosols (where organic liquids are used), and mixed sols (where the liquid medium comprises both water and an organic liquid). See, e.g., the descriptions of the techniques and forms given in U.S. Patent Nos. 2,801,185 (Iler) and 4,522,958 (Das et al.), as well as those given by R. K. Iler in The Chemistry of Silica, John Wiley & Sons, New York (1979).

Due to their surface chemistry, low cost, and environmental considerations, silica hydrosols are preferred for use in preparing the compositions of the invention. Such hydrosols are available in both acidic and basic forms and in a variety of particle sizes and concentrations from, e.g., Nycol Products, Inc. in Ashland, Maryland; Nalco Chemical Company in Oakbrook,

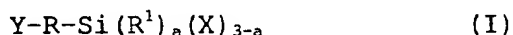
Illinois; and E. I. duPont de Nemours and Company in Wilmington, Delaware. Concentrations of from about 2 to about 50 percent by weight of silica in water are generally useful. If desired, silica hydrosols can be prepared, e.g., by partially neutralizing an aqueous solution of an alkali metal silicate with acid to a pH of about 8 or 9 (such that the resulting sodium content of the solution is less than about 1 percent by weight based on sodium oxide). Other methods of preparing silica hydrosols, e.g., electrodialysis, ion exchange of sodium silicate, hydrolysis of silicon compounds, and dissolution of elemental silicon are described by Iler, supra.

In preparing the compositions of the invention, a sol of polymer-grafted microparticles can generally be prepared first and then, if water and oil repellency are desired, combined with at least one fluorochemical. Preparation of the sol generally requires that at least a portion of the surface of the inorganic microparticles be modified by chemical reaction (or strong physical interaction) with a coupling agent. Suitable coupling agents are not only capable of reacting (or strongly interacting) with chemical groups on the surface of the microparticles, but also comprise at least one functional group, e.g., $-CH_2SH$, $-CH_2NH_2$, etc., which can function as a chain transfer site for free radical polymerization (a chain transfer group).

For example, silica microparticles can be treated with hydrolyzable, chain transfer group-containing organosilanes under conditions such that silanol groups on the surface of the particles chemically bond with hydrolyzed silane groups to produce covalent silicon-oxygen-silicon bonds. The surface of silica (or other metal oxide) particles can

also be treated with other chemical compounds, e.g., hydrolyzable, chain transfer group-containing organotitanates, which are capable of attaching to the surface of the particles by a chemical bond (covalent or ionic) or by a strong physical bond, and which
5 comprise at least one functional group, e.g., $-\text{CH}_2\text{SH}$, $-\text{CH}_2\text{NH}_2$, etc., which can function as a chain transfer site for free radical polymerization.

Treatment with hydrolyzable, chain transfer group-containing organosilanes is generally preferred.
10 A preferred class of such compounds can be represented by the general formula



15 (as well as partial hydrolyzates and condensates thereof) wherein Y is a chain transfer group (preferably, a moiety selected from the group consisting of mercapto, amino-containing, oxygen-containing, and halogen-containing groups; more preferably, mercapto); R is a divalent group
20 selected from the group consisting of alkylene groups having from 1 to about 12 carbon atoms and arylene groups having from about 6 to about 12 carbon atoms; X is a hydrolyzable group (preferably, a group selected from the group consisting of halogen, alkoxy groups having from 1
25 to about 12 carbon atoms and optionally containing from 1 to about 11 ether oxygen atoms, acyloxy, amino, and alkylamino); R^1 is selected from the group consisting of lower alkyl groups having from 1 to about 4 carbon atoms and phenyl; and a is an integer of 0 to 2 (preferably, 0).

Representative examples of preferred hydrolyzable, chain transfer group-containing organosilanes include
35 3-mercaptopropyltrimethoxysilane,

- 4-mercaptobutyltriethoxysilane,
4-mercaptobutyldimethoxymethylsilane,
6-mercaptohexyltrimethoxysilane,
4-mercaptocyclohexyltrimethoxysilane,
5 3-aminopropyltrimethoxysilane,
4-aminobutyltriethoxysilane,
4-aminobutyldimethoxymethylsilane,
6-aminoethyltrimethoxysilane,
4-aminocyclohexyltrimethoxysilane,
10 3-chloropropyltrimethoxysilane,
4-chlorobutyltriethoxysilane,
4-chlorobutyldimethoxymethylsilane,
6-chlorohexyltrimethoxysilane,
4-chlorocyclohexyltrimethoxysilane, N,N-
15 dimethylaminopropyltrimethoxysilane,
glycidoxypropyltrimethoxysilane, and mixtures thereof.

In preparing the sols of polymer-grafted microparticles, a hydrosol (e.g., a silica hydrosol), a mixed sol (e.g., a mixed silica sol, wherein the liquid
20 medium comprises an aqueous solution of at least one preferably water-soluble organic solvent such as methanol, ethanol, acetone, tetrahydrofuran, ethylene glycol, propoxyethanol, or the like), or an organosol (e.g., a silica organosol, where the organic liquid
25 medium comprises benzene, toluene, propoxyethanol, dimethyl formamide, isopropanol, ethylene glycol, or the like) can generally be first combined with coupling agent(s) and the resulting combination allowed to react. Heat (and agitation) can be applied to
30 facilitate reaction. (The skilled artisan will recognize that the concentration and/or pH of the sol may require adjustment to enhance the stability (and minimize the viscosity) of the sol during reaction with the coupling agent and, later, with monomer. Coupling
35 agent can generally be used in an amount such that at

least a portion of the surface of the microparticles is modified sufficiently to enable effective graft polymerization (upon combination with monomer and initiator).

5 Preferably, the amount of coupling agent is selected to be, e.g., from about 1×10^{-4} to about 7×10^{-3} , preferably from about 2×10^{-4} to about 3×10^{-3} , and more preferably from about 3×10^{-4} to about 2×10^{-3} millimoles of coupling agent per square meter of
10 microparticle surface area. The resulting mixture can be agitated and maintained at a temperature of, e.g., from about 20°C to about 100°C, preferably from about 50°C to about 100°C (e.g., for about one to about 24 hours) to
15 enable the reaction (or other interaction) of the coupling agent with chemical groups on the surface of the microparticles. This provides a colloidal dispersion of inorganic microparticles which have surface-attached or surface-bonded organic groups.

 One or more free-radically polymerizable,
20 ethylenically-monounsaturated monomers can then be added to the resulting colloidal dispersion of surface-modified microparticles. (Optionally, small amounts of one or more free-radically polymerizable, ethylenically-
25 polyunsaturated monomers can also be added to provide a degree of crosslinking, if desired.) Monomer(s) should generally be added in an amount greater than the molar equivalent of coupling agent utilized. For example, from about 0.001 to about 0.10 mole of monomer(s) per gram SiO_2 can be utilized. The resulting mixture (which can
30 include surfactant to enable graft polymerization of water-insoluble monomer in an aqueous medium) can be purged of oxygen, an effective amount of a thermally- or radiation-activatable free radical initiator can be added, and the mixture can then be heated (to a
35 temperature sufficient to decompose the free radical

source, e.g., from about 20°C to about 130°C) or can be irradiated to effect polymerization of the monomer.

Suitable monomers for use in preparing the polymer-grafted microparticles of the compositions of the invention are free-radically polymerizable, ethylenically-unsaturated monomers (or monomer mixtures). Useful monomers include, for example, ethylenically-unsaturated acids and anhydrides; ethylenically-unsaturated macromers; ethylenically-unsaturated, substituted and unsubstituted esters, amides, and nitriles; vinyl monomers; vinylidene monomers; other olefinic monomers such as styrene, ethylene, tetrafluoroethylene, and hexafluoropropene; fluorochemical-containing acrylates and methacrylates; and heterocyclic monomers.

Representative monomers include, for example, acrylic acid; methacrylic acid; itaconic acid; citraconic acid; aconitic acid; maleic acid; maleic anhydride; fumaric acid; crotonic acid; cinnamic acid; oleic acid; vinyl sulfonic acid; vinyl phosphonic acid; alkyl, cycloalkyl, and fluoroalkyl-containing esters of the foregoing acids, where the alkyl, cycloalkyl, and fluoroalkyl groups have from 1 to about 18 carbon atoms (such as, for example, ethyl, butyl, 2-ethylhexyl, octadecyl, 2-sulfoethyl, 3-sulfopropyl, acetoxyethyl, cyanoethyl, hydroxyethyl, and hydroxypropyl acrylates and methacrylates); amides of the foregoing acids (such as, for example, acrylamide, methacrylamide, methylolacrylamide, and 1,1-dimethylsulfoethylacrylamide); acrylate or methacrylate end-capped polyethylene glycols; acrylonitrile; methacrylonitrile; styrene; α -methylstyrene; p-hydroxystyrene; chlorostyrene; sulfostyrene; N-vinyl pyrrolidone; vinyl acetate; vinyl chloride; vinyl fluoride; vinyl ethers; vinyl sulfides; vinyl toluene;

butadiene; isoprene; chloroprene; ethylene; isobutylene; vinylidene chloride; vinylidene fluoride; tetrafluoroethylene; hexafluoropropene; sulfated castor oil; sulfated sperm oil; sulfated soybean oil; and
5 sulfonated dehydrated castor oil. Particularly useful monomers include, for example, alkyl acrylates and methacrylates having from 1 to about 18 carbon atoms, styrene, acrylic acid, methacrylic acid, itaconic acid, acrylamides and methacrylamides, fluorochemical-
10 containing acrylates and methacrylates, and sodium sulfostyrene. Monomers which are not soluble in the liquid medium can be solubilized by the addition of solubilizing organic liquid and/or surfactant in an amount sufficient to provide solution or emulsification
15 of the monomers.

Useful free radical initiators include inorganic and organic peroxides, hydroperoxides, persulfates, azo compounds, redox systems (e.g., a mixture of $K_2S_2O_8$ and $Na_2S_2O_5$), and free radical
20 photoinitiators such as those described by K.K. Dietliker in Chemistry & Technology of UV & EB Formulation for Coatings, Inks & Paints, Volume 3, pages 276-298, SITA Technology Ltd., London (1991). Representative examples include hydrogen peroxide, potassium persulfate, t-butyl
25 hydroperoxide, benzoyl peroxide, t-butyl perbenzoate, cumene hydroperoxide, and azobis(isobutyronitrile) (AIBN). The skilled artisan will recognize that the choice of initiator will depend upon the particular reaction conditions, e.g., choice of solvent.

30 Polymerization of the ethylenically-monounsaturated monomer(s) provides a colloidal dispersion of polymer-grafted microparticles. The microparticles have linear or branched, substantially uncrosslinked polymer chains covalently bonded at
35 essentially only one end to surface-attached coupling

agent. (As stated above, a degree of crosslinking (and thus multiple attachment sites) can be imparted by the inclusion of one or more ethylenically-polyunsaturated monomers, if desired.) In order to maximize the durability of coatings derived from the treatment compositions of the invention (comprising the dispersions of polymer-grafted microparticles), monomer(s) can be chosen to correspond to the surface energy of the substrate to which a treatment composition is to be applied. Thus, polar monomers can be utilized for substrates having a high surface energy, and, for substrates having a low surface energy, nonpolar monomers can be selected. Alternatively, monomers can be chosen which are capable of having specific chemical or physical interactions with the substrate.

The colloidal dispersion of polymer-grafted microparticles can be utilized (without any content or addition of fluorochemical) to impart durable dry soil resistance and, in some cases (e.g., when monomer(s) containing carboxylic acid or sulfonic acid groups are utilized), durable stain release to substrates. The dispersion can be diluted to, e.g., about 0.01 to about 20% solids (preferably, to about 0.05 to about 10% solids) in dispersing solvent for application to a substrate, or it can be concentrated (for storage or shipment) to a low solvent content or even to an essentially solventless condition (i.e., having essentially no liquid dispersing medium). Surprisingly, the solventless material can generally be essentially completely re-dispersed in solvent without evidence of aggregation.

If durable water and oil repellency are desired (in addition to durable dry soil resistance and stain release), the colloidal dispersion of polymer-grafted microparticles can be combined with at least one

fluorochemical. Optionally, the resulting mixture can be subjected to agitation and/or the application of heat, to encourage the reaction or interaction of the fluorochemical(s) with chemical groups on the surface of the microparticles. (For example, temperatures of from about 20°C to about 100°C can be maintained for periods of from one to about 24 hours.) Fluorochemical(s) can generally be added in an amount sufficient to impart the level of repellency desired for a particular application. Fluorochemical(s) which are not soluble in the liquid medium can be solubilized by the addition of solubilizing organic liquid and/or surfactant in an amount sufficient to provide solution or emulsification of the fluorochemical(s).

The resulting fluorochemical-containing treatment composition can be diluted with dispersing solvent to a concentration of, e.g., about 0.01 to about 20% solids for application to a substrate, or it can be concentrated (for storage or shipment) to a low solvent content. If desired, a two-step treatment process can be utilized, in which the colloidal dispersion of polymer-grafted microparticles is first applied to a substrate, followed by application of a composition comprising the fluorochemical(s).

Any of the well-known fluorochemicals that are capable of imparting water and oil repellency can be used in the compositions of the invention. Preferably, the fluorochemical(s) will be capable of chemical reaction or strong physical interaction with chemical groups on the surface of the microparticles to enhance durability, but fluorochemical(s) which do not possess such capabilities can also be utilized.

Suitable blendable fluorochemicals include any of the fluorochemical group-containing polymeric and oligomeric compounds known in the art to impart

water and oil repellency to substrates. These polymeric and oligomeric fluorochemical treatments typically comprise one or more fluorochemical groups that contain a perfluorinated carbon chain having from
5 3 to about 20 carbon atoms, more preferably from about 6 to about 14 carbon atoms. These fluorochemical groups can contain straight chain, branched chain, or cyclic fluorinated alkylene groups or any combination thereof. The fluorochemical groups are preferably free
10 of polymerizable olefinic unsaturation but can optionally contain catenary (i.e., in-chain, bonded only to carbon) heteroatoms such as oxygen, divalent or hexavalent sulfur, or nitrogen. Fully-fluorinated groups are preferred, but hydrogen or chlorine atoms
15 can also be present as substituents, provided that no more than one atom of either is present for every two carbon atoms. It is additionally preferred that any fluorochemical group contain from about 40% to about 80% fluorine by weight, more preferably about 50% to
20 about 78% fluorine by weight. The terminal portion of the group is generally fully-fluorinated, preferably containing at least 7 fluorine atoms, e.g., $\text{CF}_3\text{CF}_2\text{CF}_2-$, $(\text{CF}_3)_2\text{CF}-$, SF_5CF_2- . Perfluorinated aliphatic groups (i.e., those of the formula $\text{C}_n\text{F}_{2n+1}-$) are the most
25 preferred fluorochemical groups.

Representative examples of suitable blendable fluorochemicals include fluorochemical urethanes, ureas, esters, ethers, alcohols, epoxides, allophanates, amides, amines (and salts thereof), acids
30 (and salts thereof), carbodiimides, guanidines, oxazolidinones, isocyanurates, biurets, acrylate and methacrylate homopolymers and copolymers, and mixtures thereof.

Representative blendable fluorochemical
35 group-containing polymers useful in the present

invention include fluorochemical acrylate and methacrylate homopolymers or copolymers containing fluorochemical acrylate monomers interpolymerized with monomers such as methyl methacrylate, butyl acrylate, octadecylmethacrylate, acrylate and methacrylate esters of oxyalkylene and polyoxyalkylene polyol oligomers (e.g., diethylene glycol dimethacrylate, polyethylene glycol dimethacrylate, polyethylene oxide diacrylate, and polyethylene glycol monoacrylate), glycidyl methacrylate, ethylene, butadiene, styrene, isoprene, chloroprene, vinyl acetate, vinyl chloride, vinylidene chloride, vinylidene fluoride, acrylonitrile, vinyl chloroacetate, vinylpyridine, vinyl alkyl ethers, vinyl alkyl ketones, acrylic acid, methacrylic acid, 2-hydroxyethylacrylate, N-methylolacrylamide, 2-(N,N,N-trimethylammonium)ethyl methacrylate, and 2-acrylamido-2-methylpropanesulfonic acid (AMPS). The relative amounts of various comonomers used can generally be selected empirically, depending on the substrate to be treated, the properties desired, and the mode of application to the substrate. Useful fluorochemical treatments also include blends of the various fluorochemicals described above.

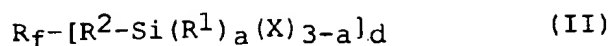
Also useful in the present invention are blends of fluorochemicals with fluorine-free extender compounds, such as siloxanes, (meth)acrylate and substituted acrylate polymers and copolymers, N-methylolacrylamide-containing acrylate polymers, urethanes, blocked isocyanate-containing polymers and oligomers, condensates or precondensates of urea or melamine with formaldehyde, glyoxal resins, condensates of fatty acids with melamine or urea derivatives, condensates of fatty acids with polyamides and their epichlorohydrin adducts, waxes, polyethylene, chlorinated polyethylene, alkyl ketene dimers, esters,

and amides. Blends of these fluorine-free extender compounds can also be used. The relative amount of extender compound to fluorochemical is not critical. However, the overall composition of the fluorochemical treatment generally contains, relative to the amount of solids present in the system, at least about 3 weight percent, preferably at least about 5 weight percent, carbon-bound fluorine in the form of said fluorochemical groups.

Many fluorochemicals, including blends that include fluorine-free extender molecules such as those described above, are commercially available as ready-made formulations. Such products are sold, for example, as Scotchgard™ brand Carpet Protector (manufactured by 3M Co., Saint Paul, Minnesota) and as Zonyl™ brand Carpet Treatment (manufactured by E.I. du Pont de Nemours and Company, Wilmington, Delaware).

Useful fluorochemicals are described in U.S. Patent Nos. 3,728,151 (Sherman et al.), 3,816,229 (Bierbrauer), 3,896,035 (Schultz et al.), 3,901,727 (Loudas), 3,916,053 (Sherman et al.), 4,043,923 (Loudas), 4,043,964 (Sherman et al.), 4,264,484 (Patel), 4,624,889 (Bries), and 5,274,159 (Pellerite et al.).

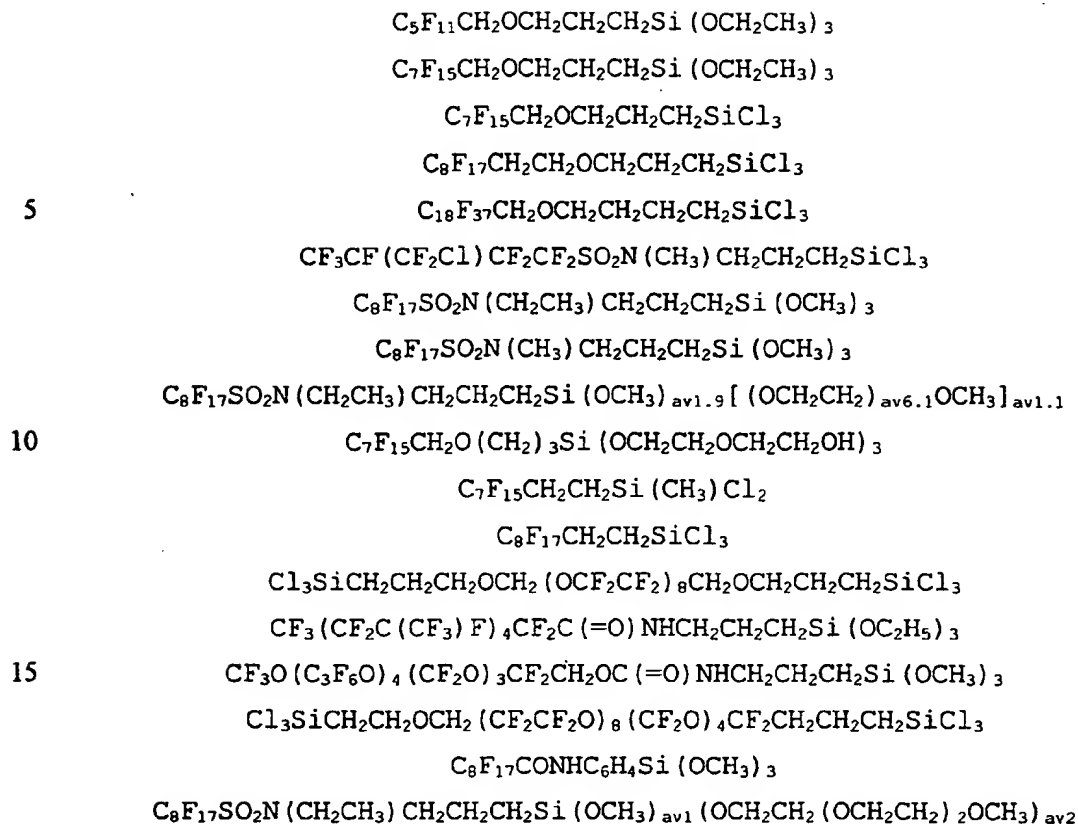
Preferred fluorochemicals for use in the compositions of the invention are those which are capable of chemical reaction or strong physical interaction with chemical groups on the surface of the microparticles. Most preferred are fluorochemical silanes having at least one hydrolyzable or hydrolyzed group. These are compounds, mixtures of compounds, polymers, or copolymers containing at least one silicon atom bonded to at least one group that is capable of reacting with, e.g., silanol groups present on the surface of silica microparticles. A class of useful fluorochemical silanes can be represented by the following general formula:



(as well as partial hydrolyzates and condensates thereof)
 5 where R^1 , X , and a are as defined above for formula I; R_f
 is a monovalent or divalent, fluoroaliphatic or
 fluoroalicyclic group containing at least about 4 carbon
 atoms and at least about 25 (preferably about 50) weight
 percent fluorine; d is an integer of 1 or 2 (preferably
 10 1); and R^2 is a divalent linking group which preferably
 comprises at least one group selected from the group
 consisting of arylene (preferably phenylene), and
 alkylene of 1 to about 12 carbon atoms, which may further
 comprise one or more divalent groups that are
 15 substantially stable against hydrolysis (e.g., $-CO-$,
 $-CONR^3-$, $-SO_2-$, $-SO_2NR^3-$, $-(C_2H_4O)_b-$, and $-(C_3H_6O)_b-$, where b
 is an integer of 1 to about 4 and R^3 is hydrogen or alkyl
 of 1 to about 4 carbon atoms).

Preferably, R_f is a linear or branched
 20 fluoroaliphatic group terminated with a trifluoromethyl
 group and having from about 4 to about 24 carbon atoms
 (more preferably, from about 4 to about 12) or is a
 fluoroalicyclic group having from about 4 to about 24
 carbon atoms. R_f can optionally contain catenary ether
 25 oxygen atoms (bonded only to carbon atoms) and can
 contain some hydrogen or chlorine, provided that not more
 than one atom of either hydrogen or chlorine is present
 for every two carbon atoms. Most preferably, R_f is a
 linear or branched, perfluoroaliphatic or
 30 perfluoroalicyclic group having from about 4 to about 12
 carbon atoms.

Representative examples of useful
 fluorochemical silanes include the following compounds:



20 The fluorochemical silanes can be mixed with other non-fluorinated silanes, if desired. Partial condensates of such mixtures can also be utilized.

Useful fluorochemical silanes can be prepared, e.g., by combining at least one fluorochemical compound having at least one reactive functional group with a
 25 silane having from one to about three hydrolyzable groups and at least one alkyl, aryl, or alkoxyalkyl group that is substituted by at least one functional group that is capable of reacting with the functional group of the fluorochemical compound(s). Suitable fluorochemical
 30 compounds can be prepared, e.g., by the methods disclosed in U.S. Patent No. 5,274,159 (Pellerite et al.).

The treatment compositions of the invention can be used to impart durable dry soil resistance, in some cases durable stain release, and (if a fluorochemical-
 35 containing composition is utilized) durable water and oil

repellency to substrates such as carpets, fabrics (both woven and non-woven), papers, leather, plastics, wood, metals, glass, ceramics, stone, and cement. The compositions can generally be diluted (e.g., to a solids concentration of about 0.01 to about 20 weight percent) and applied to a substrate by any of the customary procedures known in the art, e.g., by padding, spraying, exhaustion, knife-coating, dipping, roll-coating, foam techniques, and the like. (When treating nylon substrates by an exhaustion technique, the pH of the compositions can be adjusted to about 1.5 to about 3 using a nonvolatile acid, and a salting agent can be added.) Other conventional additives, e.g., wetting agents, antistats, stain release agents, biocides, antioxidants, and the like, can be included in the compositions of the invention.

If desired, fiber or yarn substrates can be treated prior to conversion to a finished product such as a carpet or rug. Treated substrates can be allowed to dry at ambient temperature or at elevated temperatures.

The quantity of treatment composition employed can vary within wide limits, depending upon the nature of the substrate, the nature of the components of the composition, and the properties desired. The amount should be sufficient to provide the desired degree of stain release and antisoiling characteristics or these characteristics coupled with oil- and water-repelling characteristics.

Objects and advantages of this invention are further illustrated by the following examples, but the particular materials and amounts thereof recited in these examples, as well as other conditions and details, should not be construed to unduly limit this invention. In the examples, all temperatures are in

degrees Centigrade and all parts and percentages are by weight unless indicated otherwise.

Examples

5

Test Procedures

Stain Release (SR)

10 Stain release properties of a carpet sample
were characterized by introducing 20 ml of cherry
flavored KOOL-AID™ Soft Drink Mix (available from Kraft
General Foods, Northfield, Illinois) (or a 0.1 wt %
FD&C Red Dye # 40 solution adjusted to pH 3.2 with
15 citric acid) challenge solution into an open ended
cylinder (5 cm in height and 5 cm in diameter) placed
on the tufted surface of a carpet test sample. If the
challenge solution was initially repelled by the
carpet, the cylinder or the carpet was tapped until the
fluid penetrated into the carpet pile. The cylinder
20 was removed, the stained carpet sample allowed to sit
at room temperature (about 22°C) for 24 hours, after
which the carpet sample was rinsed with water until no
color was present in the effluent. The carpet sample
was then centrifuged to remove most of the water and
25 dried at 65°C. Residual stain (i.e. the degree of
reddening) on the dried carpet was measured using a
Minolta CR-310 Light Meter, comparing the stained value
to an unstained control sample of the same carpet with
the stain release treatment. The degree of staining is
30 reported as Δa (the difference between the stained and
unstained samples), with the lower Δa values
representing better stain release, or less residual
staining of the carpet. (Based on a three dimensional
CIE $L^*a^*b^*$ color scale, where a^* represents the red-

green vector, L* the light/dark ratio, and b* the blue/yellow ratio.)

Dynamic Water Resistance (DWR)

- 5 Dynamic water resistance of a carpet sample was determined by inclining a test sample (15.2 cm X 30.5 cm) at an angle of 45° from horizontal and impinging deionized (D.I.) water (20 ml) onto the center of the carpet sample through a 5 mm i.d. tube positioned 45.7 cm above the sample. The increase in weight of the test sample provided an indication of its ability to repel water, with lower weight gains indicating better repellency properties.

15 Oil/Aqueous Repellency (OR/AR)

- The ability of a carpet sample to repel oil challenges (OR) was determined using the American Association of Textile Chemists and Colorists (AATCC) Standard Test Method No. 118-1983, which test is based on the resistance of a treated fabric to penetration by oils of varying surface tensions. Oils and their associated rating numbers include:

AATCC Oil Repellency Rating Number	Oil
1	Mineral Oil
C	85:15 Mineral Oil/Hexadecane (v/v)
2	65:35 Mineral Oil/Hexadecane (v/v)
3	n-Hexadecane
4	n-Tetradecane
5	n-Dodecane
6	n-Decane
7	n-Octane
8	n-Heptane

Five drops of each oil challenge were placed on a carpet sample and the number of drops of each challenge remaining on the carpet surface (i.e. not soaked into the carpet pile) after one minute noted.

5 An "Oil Repellency" number was assigned based on the highest rating number challenge (i.e. lowest surface tension) of which all five test drops did not soak into the carpet sample. The higher the oil rating number, the better the oil repellency properties of the carpet.

10 The ability of a carpet sample to repel aqueous challenges (AR) was determined with water/isopropyl alcohol test challenges. Test solutions and their rating numbers included:

Aqueous Challenge Fluid Rating Number	Fluid
0	Water
1	90:10 Water/Isopropyl alcohol (v/v)
2	80:20 Water/Isopropyl alcohol (v/v)
3	70:30 Water/Isopropyl alcohol (v/v)
4	60:40 Water/Isopropyl alcohol (v/v)
5	50:50 Water/Isopropyl alcohol (v/v)
6	40:60 Water/Isopropyl alcohol (v/v)
7	30:70 Water/Isopropyl alcohol (v/v)
8	20:80 Water/Isopropyl alcohol (v/v)
9	10:90 Water/Isopropyl alcohol (v/v)
10	Isopropyl alcohol

15

Five drops of each aqueous challenge fluid were placed on a carpet sample and the number of drops of each fluid remaining on the carpet surface (i.e., not soaking into the carpet pile) after one minute noted. An "Aqueous Repellency" number was assigned

20

based on the highest rating number fluid (i.e., lowest surface tension) of which all five test drops did not soak into the carpet sample. The higher the aqueous rating number, the better the aqueous repellency properties of the carpet.

Elemental Si Analysis

Elemental Si content of treated carpet samples was determined by obtaining a solution of Si containing treated carpet fibers and analyzing the solution for Si by Inductively Coupled Plasma (ICP) analysis.

Sulfuric acid (15 ml) was added to approximately 1 g of treated carpet fibers and the mixture placed on a warm hot plate until the fibers charred. Nitric acid was added to the charred mixture until the mixture turned clear yellow, at which point perchloric acid (3 ml) was added to obtain a colorless sample. The sample was then evaporated to approximately 5 ml, cooled, and the container rinsed with D.I. water. The acid solution and wash were combined and transferred to a plastic volumetric flask (100 ml), hydrofluoric acid (3 ml) added to the solution and D.I. water added to dilute the sample to 100 ml. The sample was then analyzed against Si standards with the same acid matrix using a Thermo Jarrell Ash ICAP 61E spectrophotometer. The Si content was translated to % Solids on Carpet (SOC) of the SiO₂/methacrylic acid (MAA) graft based on the assumption that 1 mole Si corresponds to 1 mole SiO₂ and all MAA was grafted to silica during the graft polymerization reaction. Si content is reported as ppm Si. In all cases, untreated carpet samples were included in the analysis to verify that other sources of elemental Si (e.g., from silicone oils) were not

contributing to the Si content on the treated carpet samples.

Durability Simulation

5 The durability of carpet treatments toward
"steam cleaning" was characterized by soaking a treated
carpet in a pH 10 solution of DUPONOL WAQE (a sodium
lauryl sulfate based surfactant system, available from
Witco Corp., Greenwich , CT) under conditions which
10 simulated the effect of multiple "steam cleanings."
The surfactant was present at a concentration of 7 ml
per liter water, and the solution pH was maintained at
10 with trisodium phosphate. Following the soak, the
carpet sample was extensively rinsed with tap water,
15 centrifuged to remove most of the water, and dried at
room temperature. The thus-challenged carpet samples
were tested for Stain Release, Dynamic Water
Repellency, and silica content as described above, the
results of the tests being reported as WAQE SR, WAQE
20 DWR, and WAQE ppm Si respectively. Carpet treatments
that retain at least about 25% of their original
performance after the "WAQE" challenge show reasonable
durability toward steam cleaning.

25 Walk-on Soiling

Treated and untreated (control) carpet
squares were mounted on particle board, the mounted
samples placed in a commercial location, and the carpet
samples soiled by normal foot traffic. The amount of
30 foot traffic was monitored and the position of each
test sample within the test location changed daily
using a pattern designed to minimize the effects of the
carpet sample position and orientation on soiling.
Following a specific soil challenge period, measured in
35 numbers of cycles wherein one cycle equals

approximately 10,000 foot traffics, the samples were removed and vacuumed to remove unadhered debris. The amount of soil remaining on each sample was determined using colorimetric measurements, with the assumption
5 that the amount of soil on a sample was proportional to the difference in color between the unsoiled sample and the corresponding sample after soiling. The three CIE $L^*a^*b^*$ color coordinates of the unsoiled and subsequently soiled samples were measured using a
10 Minolta CR-310 Chroma Meter with a D65 illumination source. The color difference value, ΔE , was calculated using the following equation:

$$\Delta E = [(\Delta L^*)^2 + (\Delta a^*)^2 + (\Delta b^*)^2]^{1/2}$$

15 where: $\Delta L^* = L^*_{\text{soiled}} - L^*_{\text{unsoiled}}$
 $\Delta a^* = a^*_{\text{soiled}} - a^*_{\text{unsoiled}}$
 $\Delta b^* = b^*_{\text{soiled}} - b^*_{\text{unsoiled}}$

ΔE values calculated from these colorimetric
20 measurements have been shown to be qualitatively in agreement with values from older, visual evaluations such as the soiling evaluation suggested by the AATCC, but possess the additional advantages of having a degree of precision and of being unaffected by
25 evaluation environment or operator. Final ΔE values for each sample were calculated as an average of between five and seven replicates.

The performance improvement of a sample above the untreated control sample, reported as $\Delta\Delta E$, was
30 calculated using the equation below:

$$\Delta\Delta E = \Delta E_{\text{soiled untreated control}} - \Delta E_{\text{soiled treated}}$$

Coupling Agent Treated Silica Precursor Preparation

Silica sols having chain transfer coupling agent treatments were prepared according to the following general procedure:

- 5 Coupling agent addition to the silica sol suspension was typically carried out at a silica sol concentration of 5-20% solids. Sols available at concentrations higher than 20% solids were typically diluted with D.I. water to obtain a more workable
- 10 concentration (i.e., a concentration which would not gel or become too viscous during the coupling agent treatment or subsequent graft polymerization). The pH of the suspension was adjusted to about 2.5-3.5 by the addition of acetic acid, sulfamic acid, or sulfuric
- 15 acid prior to the addition of the coupling agent. Silane coupling agent, as indicated in Table 1, was added to the acidic suspension and the resulting mixture stirred at 75-80°C for 16-24 hours. (Initially
- 20 the silane coupling agent formed an oily film on the surface of the aqueous suspension but after the 16-24 hour reaction the mixture was homogeneous.) The amount of silane coupling agent used in this reaction was adjusted to correspond to the mmol coupling agent/g silica sol ratio indicated in Table 1. Subsequent to
- 25 the completion of the reaction, the reaction mixture was diluted with additional D.I. water to produce an approximately 2.5-7.5% solids, preferably about 3% solids, suspension of the coupling agent treated silica sols.

Table 1
Coupling Agent Treated Precursors

Precursor #	Silica	Coupling Agent	mmol Coupling Agent/g Silica
P-1	NALCOAG™ 1042 ¹	Mercaptopropyl-trimethoxysilane ⁶	0.05
P-2	NALCOAG™ 1042	Mercaptopropyl-trimethoxysilane	0.25
P-3	NALCOAG™ 1042	Glycidoxypropyl-trimethoxysilane ⁷	0.05
P-4	NALCOAG™ 1056 ²	Mercaptopropyl-trimethoxysilane	0.05
P-5	NALCOAG™ 2326 ³	Mercaptopropyl-trimethoxysilane	0.05
P-6	NALCOAG™ 2329 ⁴	Mercaptopropyl-trimethoxysilane	0.014
P-7	CAB-O-SPERSE™ S 3295 ⁵	Mercaptopropyl-trimethoxysilane	0.10

1. A colloidal SiO₂ suspension having a mean particle size of 20 nm, available from Nalco Chemical Co., Chicago, IL, as a 34% solids suspension in water at pH 3.2 and a viscosity of <10 cp.
2. A colloidal SiO₂ core with an Al₂O₃ shell (~4% Al₂O₃) suspension having a mean particle size of 20 nm, available from Nalco Chemical Co. as a 30% solids suspension in water at pH 4.2 and a viscosity of 15 cp.
3. A colloidal SiO₂ suspension having a mean particle size of 5 nm, available from Nalco Chemical Co. as a 15% solids suspension in water at pH 9.2 and a viscosity of 5 cp.
4. A colloidal SiO₂ suspension having a mean particle size of 75 nm, available from Nalco Chemical Co. as a 30% solids suspension in water at pH 9.2 and a viscosity of 10 cp.
5. An amorphous fumed silica available from Cabot Corp., Cab-O-Sil Division, Tuscola, IL as a 15% solids aqueous slurry at pH 9.5-10.
6. Mercaptopropyltrimethoxysilane (MPTMS), available from Aldrich Chemical Co., Milwaukee, WI.
7. Glycidoxypropyltrimethoxysilane (GPTMS), available from Aldrich Chemical Co.

Examples 1 - 10

Polymer grafted silica sols were prepared according to the following general procedure:

Monomer(s) in the silica/monomer ratios indicated in Table 2 were added to approximately 3% solids suspensions of the indicated precursor, the reaction mixture deoxygenated by a N₂ purge for about 5 minutes and then heated to 65-70°C under a N₂ atmosphere, and initiator (e.g., potassium persulfate or t-butyl hydroperoxide) added to the reaction mixture, typically at an approximately 1% to monomer concentration. The reaction mixture was maintained at 65-70°C under a N₂ atmosphere, with stirring, for 24 hours to complete the graft polymerization. The resulting clear, bluish suspension was free from the smell of monomer and was used without further purification.

Table 2
Polymer Grafted Silica Sols

Example #	Precursor #	Monomer(s)	Silica:Monomer Ratio (wt:wt)
1	P-1	Methacrylic Acid ¹	1:1
2	P-1	Methacrylic Acid	1:1.5
3	P-1	Methacrylic Acid	1:2
3a	P-1	Methacrylic Acid	2:1
4	P-2	Methacrylic Acid	1:1
5	P-3	Methacrylic Acid	1:1
6	P-4	Methacrylic Acid	1:1
7	P-5	Methacrylic Acid	1:1
8	P-6	Methacrylic Acid	1:1
9	P-1	Methacrylic Acid: Butyl acrylate ² (83:17) ³	1: 1.7
C -1 ⁴	P-7	Methacrylic Acid	1:1

1. Methacrylic Acid (MAA), available from Aldrich Chemical Co. Inhibitor was removed from the MAA just prior to the graft polymerization by passing it through a column of De-Hibit-100 macroreticular ion exchange resin (available from Polysciences, Inc., Warrington, VA, in the form of water-saturated chloride beads) or by distillation.

2. Butyl Acrylate (BA), available from Aldrich Chemical Co. Inhibitor was removed from the BuA in the same manner it was removed from MAA.
3. This reaction mixture also contained 3MTM Stain Release Concentrate FC-369 (11 parts solids relative to the total of MAA, SiO₂, and BuA), a sulfonated Novalak resin available from 3M Co., St. Paul, MN.
4. This example is included as a comparative example where the silica particles comprise agglomerations of smaller particles rather than the substantially discrete particles of the present invention.

Comparative Example C-2

D.I. water (450 g) was added to NALCOAGTM 1042 (36.75 g) to produce an approximately 2.5% solids colloidal suspension. Freshly distilled methacrylic acid (12.5 g) was added to the silica suspension, the resulting mixture deoxygenated by a N₂ purge for about 5 minutes, the mixture heated to 80°C under a N₂ atmosphere, and t-butyl hydroperoxide (1.5 g of a 70% actives aqueous solution, available from Aldrich Chemical Co.) added to the mixture. The resulting mixture was maintained at 80°C under a N₂ atmosphere, with stirring, for 24 hours to complete polymerization. The resulting clear, bluish suspension was free from the smell of monomer and was used without further purification.

Carpet Treatment Procedures

The polymer grafted silica sols of the present invention were applied to various nylon 6 or nylon 66 carpets by the following exhaustion or flexnip procedures.

Exhaustion Application (EX)

A 1000 ml stainless steel beaker was charged with 500 g of a 1% solids (wt) aqueous dispersion of the composition to be exhausted onto the carpet sample.

Optionally, MgSO_4 (0.5 wt %) and/or Karawet™ DOSS (0.1 wt %; dioctylsulfosuccinate, available from Rhone Poulenc, Baltimore, MD) was added to the dispersion. The pH of the dispersion was adjusted to 2 with a 10 wt % solution of sulfamic acid and the beaker then partially immersed in an oil bath at 100°C. When the temperature of the dispersion reached 76°C, an approximately 10 g sample of a nylon carpet (a light blue nylon 6,6 carpet having a face weight of about 1.2 kg/m²) was completely immersed in the dispersion and the carpet agitated in the suspension for 15 minutes. The carpet sample was subsequently removed from the bath and exhaustively rinsed in D.I. water. Cloth towels were applied to the sample with hand pressure to absorb most of the water from the carpet. At this point, the sample was either dried at room temperature and submitted for Si analysis or sprayed with an aqueous solution of a water soluble fluorocarbylalkoxysilane (FCSi) (prepared essentially as described in Example 6 of U.S. Pat. No. 5,274,159 (Pellerite et al.)), as described below.

Flexnip Application (FN)

Weighed carpet samples (about 15 cm X 30 cm) were soaked in D.I. water for at least 5 minutes before being subjected to forced steam in a Mathis Laboratory Steamer Type DH (available from Werner Mathis U.S.A., Inc., Concord, NJ) at steam temperature of approximately 100°C for 2 minutes with the tuft side of the carpet sample facing upward as the sample passed through the steamer. The steamed carpet was centrifuged to 10-15% wet pickup weight based on the dry weight of the carpet and the samples stored in a

sealed plastic bag (up to a maximum of 3 hours) until a treatment was applied.

The treatment bath was sized so that the carpet sample would absorb the total bath (typically 4-5 times the weight of the dry carpet sample). The desired % Solids on Carpet (%SOC) was calculated based on the carpet weight and the appropriate concentrations of treatment components (e.g., grafted silica sol, wetting agents, etc.) added to D.I. water, the bath diluted to the desired weight, and the pH of the bath adjusted to 2 with a 10% aqueous solution of sulfamic acid. Typically, 1% SOC of the grafted silica was mixed in the bath along with 0.5% SOC of MgSO_4 . (For example, to obtain a 1% SOC treatment on a 100g carpet sample, 1 g of grafted silica solids and 0.5 g MgSO_4 were added to the water for a total bath weight of 500 g). The bath was poured into a 6"x12" pan and the steamed carpet immersed tuft side down in the bath for 1 minute with a weight placed on the carpet sample to facilitate fluid pickup. The carpet was then placed tuft side down in the steamer, any remaining bath poured on top of the carpet, and the carpet steamed a second time using the same conditions as above. Immediately following steaming, the carpet was dipped in approximately 2 liters of D.I. water three times, then centrifuged to 10-15% wet pickup based on the dry weight of the carpet.

Fluorochemical treatments were applied to the treated carpets immediately following the flexnip treatment or, alternatively, the wet treated samples were stored in sealed plastic bags up to a maximum of 3 hours between the flexnip and spray operations.

Fluorochemical Component Application

5 The fluorochemical component of the carpet
treatment compositions was applied to damp carpet
samples following the exhaustion or flexnup application
of the grafted silica component. An aqueous solution
of the fluorochemical component, with solids adjusted
so as to produce approximately 500ppm elemental F
pickup, based on the dry weight of the sample, with a
15% wet weight add-on, was applied to the damp carpet
10 test sample as a topical spray. The carpet samples
were dried at 121°C and allowed to stand at room
temperature for at least 8 hours before testing.

Examples 10-40 and 46-54

15 Carpet samples were prepared using the
grafted silica and grafted silica/fluorochemical
compositions of the present invention and the
application techniques described above, as indicated in
Table 3.

20

Table 3
Carpet Sample Preparation

Example #	Carpet Type	Grafted Si Example #	Application Process	% Solids on Carpet	Fluorochemical Component
Control-1 ¹	Carpet 1 ²	-	-	-	-
10	Carpet 1	1	FN	1.0	none
11	Carpet 1	1	FN	1.0	FCSi ³
Control-2 ⁴	Carpet 1	-	-	-	-
12	Carpet 1	1	FN	1.0	FC-1373M ⁴
13	Carpet 1	1	FN	1.0	FC-1355 ⁵
14	Carpet 1	9	FN	1.0	FC-1373M
15	Carpet 1	9	FN	1.0	FC-1355
Control-3 ¹	Carpet 1	-	-	-	-
16	Carpet 1	1	FN	0.2	FCSi
17	Carpet 1	7	FN	1.0	FCSi
18	Carpet 1	7	FN	0.2	FCSi
19	Carpet 2 ⁶	1	FN	1.0	FCSi
20	Carpet 3 ⁷	1	FN	1.0	FCSi
21	Carpet 2	2	FN	1.0	FCSi
22	Carpet 3	2	FN	1.0	FCSi
23	Carpet 2	2	FN	1.5	FCSi
24	Carpet 3	2	FN	1.5	FCSi
25	Carpet 2	3	FN	1.0	FCSi
26	Carpet 3	3	FN	1.0	FCSi
27	Carpet 2	3	FN	2.0	FCSi
28	Carpet 3	3	FN	2.0	FCSi
29	Carpet 2	4	FN	1.0	FCSi
30	Carpet 3	4	FN	1.0	FCSi
31	Carpet 2	6	FN	1.0	FCSi
32	Carpet 1	6	FN	1.0	FCSi
33	Carpet 2	8	FN	1.0	FCSi
34	Carpet 1	8	FN	1.0	FCSi
35	Carpet 3	8	FN	1.0	FCSi
Comp. C-3 ⁸	Carpet 1	C-1	FN	1.0	FCSi
Comp. C-4 ⁸	Carpet 4 ⁹	C-1	EX	-	-
36	Carpet 4	5	EX	-	-
Comp. C-5 ¹⁰	Carpet 4	-	-	-	-
37	Carpet 4	1	EX	-	none
Comp. C-6	Carpet 4	C-2	EX	-	none
Control-4 ¹	Carpet 4	-	-	-	-
38	Carpet 4	3 ¹¹	EX	-	none
40	Carpet 4	39	EX	-	none
46	Carpet 1	3	FN	1.0	FCSi
47	Carpet 1	3	FN	1.5	FCSi
48	Carpet 1	3a	FN	0.75	FCSi
49	Carpet 1	3a	FN	1.0	FCSi
50	Carpet 1	4	FN	1.0	FCSi
51	Carpet 3	3a	FN	0.75	FCSi
52	Carpet 3	3a	FN	1.0	FCSi
53	Carpet 3	3	FN	1.5	FCSi
54	Carpet 3	6	FN	1.0	FCSi

1. Untreated sample of carpet.
2. An off-white colored nylon 66 carpet having a face weight of about 1.3 kg/m². Used for walk-on soiling evaluations because it provided good contrast between soiled and unsoiled areas and consistent reproducibility between tests.
3. A fluorocarbylalkoxysilane prepared essentially as described in Example 6 of U.S. Pat. No. 5,274,159 (Pellerite et. al.).
4. Scotchgard™ FC-1373M Commercial Carpet Protector, a 31% (wt) solids aqueous treatment containing a fluorochemical urethane, available from 3M Co.
5. Scotchgard™ FC-1355 Commercial Carpet Protector, an aqueous fluoroaliphatic polymer treatment containing 45.6% solids, available from 3M Co.
6. A light blue colored nylon 6 carpet having a face weight of about 1.2 kg/m² used for stain resistance evaluations.
7. A light blue colored nylon 66 carpet having a face weight of about 1.4 kg/m² used for oil and water repellency evaluations.
8. Silica particles used in this treatment were agglomerates of smaller silica particles rather than the substantially discrete particles of the present invention.
9. A light blue nylon 6,6 carpet having a face weight of about 1.2 kg/m².
10. NALCOAG™ 1042 (1% SiO₂ solids in bath) exhausted onto Carpet 4.
11. The treatment suspension of Example 3 was evaporated to dryness in a rotary evaporator and redispersed in water by the same procedure used to redisperse the silica sol produced in Example 39.

35

The performance characteristics of the protective compositions of the present invention on the above prepared carpet samples were characterized using the above described test methods, the results of which are reported in Table 4.

40

Performance data for the protected carpets described in Examples 10-40 and 46-54 has been grouped according to carpet type and the specific performance feature being tested (i.e., Carpet 1 was used for walk on soiling evaluations, Carpet 2 was used for stain resistance evaluations, Carpet 3 was used for oil and water repellency evaluations, and Carpet 4 was used to determine Si levels before and after exhaustion treatment).

Table 4
PROTECTIVE TREATMENT PERFORMANCE

Example #	Walk-On Soiling (ΔA_E)	SR (Δa)	WAOE SR (Δa)	DWR (g)	WAOE DWR (g)	K ₁ OR/AR	ppm Si	WAOB ppm Si
10	3.69							
11	4.87	15.6	35.9	5.1	10.9	2/2	1260	831
12	4.58	11.9	27.7				1382	1385
13	3.51	16.4	32.4	13.8	19.5		1375	1170
14	2.87	23.1	30.9				893	693
15	3.48	26.4	32.3	10.8	13.5		1069	870
16	1.72					3/4		
46	4.60					C/1	679	297
47	5.97					C/1	1492	671
48	5.1					C/1	1343	1088
49	4.78					1/1	1791	895
50	2.95					2/1	422	135
17	3.09			6.3		3/2		
18	1.5			4.8		4/5		
32	4.35	18.4	37	4.4		3/1	497	329
34	4.42	36.8	38.2	4.9		2/1	1191	497
Comp. C-3		33.8	42.4	5.2	17.2	3/2	270	234
19		5.5	23.7				1037	755
21		1.2	20.7				708	285
23		1.3	7.1					
25		1.1	13				415	198
27		0.8	1.4				1689	1097
29		2.4	10.8	7.4	17.8	3/2	1337	679
31		1.5	9.7	3.4		3/F†		

Table 4 (continued)
PROTECTIVE TREATMENT PERFORMANCE

Example #	Walk-On Soling (Δa)	SR (Δa)	WAGE SR (Δa)	DWR (g)	WAGE DWR (g)	KH OR/AR	ppm Si	WAGE ppm Si
33		15.5	17.8	3.8		3/F†		
20		15.24	22	3.5	5.3	C/2	897	727
51				3.9	6.7	C/1	1823	1595
52				3.5	5.3	C/1	1463	1179
22		7.6	19.8	3.7			807	
24		12.9	19.9	3.8			640	
26		13	20.8	3.5	15.1	1/1	425	238
53				5.6	14.7	2/1	1746	1421
28		16.1	16.2	5.8			1069	871
30		12.3	23.6	4.1	10.9	C/2	1153	948
35		12.6	17.3	3.5	14.5	1/1	1271	1026
54				3	4.7	3/1	354	351
36							507	
37							4957	
38							5181	
40							916	
Control-1	0*							
Control-2	0*	14.4						
Control-3	0*			18.6		F/F†		
Control-4							154	
Comp. C-4							292	
Comp. C-5							906	
Comp. C-6							65	

* By definition, Δa for controls is 0.

† F = No observed repellency to any challenge fluid.

Example 39

Methacrylic acid dried over 5Å molecular sieves and free from inhibitor (22.4 g), 3-mercaptopropyltrimethoxysilane (MPTMS; 0.56 g), and initiator (0.23 g, azobis(isobutyronitrile) (AIBN), available from Aldrich Chemical Co.) were added to freshly dried methanol (198 g; distilled over CaH₂) and the mixture deoxygenated by a N₂ purge for approximately 10 minutes. The reaction mixture was then heated to reflux, with stirring and under a N₂ blanket, for 24

hours. NALCOAG™ 1042 (29.4 g) was added to 50 g of the methanol polymer solution and the reaction mixture heated at reflux for 24 hours. The resulting mixture was evaporated to dryness in a rotary evaporator to produce a light yellow solid having a 0.06 mmol MPTMS/g SiO₂ ratio and a 2:1 SiO₂:MAA ratio.

5 g of the solid was added to 495 g D.I. water and sufficient HO₃SNH₂ added to adjust the pH to 2 and the solution used to treat carpet samples via the previously described exhaustion application technique.

Example 41

A treatment solution was prepared by adjusting 500 g of a 2 wt. percent solids suspension of the grafted silica of Example 1 to pH 6 using NH₄OH and mixing the suspension with a solution of 10 g of FCSi in 490 g D.I. water to produce a clear suspension. A portion of the suspension was applied to a wooden tongue depressor and rubbed into the wood using finger pressure and the depressor subsequently dried at 140°C for 3 minutes. After cooling to ambient temperature, the depressor demonstrated excellent dynamic repellency in a stream of water impinging on the depressor from a height of approximately 30 cm. Oil repellency testing on the treated depressor demonstrated beading through oil # 3 and hold out through oil # 6 after 30 seconds.

The treatment suspension was applied to a second depressor in a similar manner, but the depressor was allowed to dry at ambient temperature over night. The second depressor also demonstrated good dynamic water repellency.

Example 42

NALCOAG™ TX-8800 (a colloidal silica dispersion having a mean particle size of 20 nm,

available from Nalco Chemical Co. as a 29% solids suspension in isopropanol) was mixed with 3-mercaptopropyltrimethoxysilane (1.423 g, 0.25 mmol/g silica) and the resulting mixture heated, with stirring, at 65°C overnight. 17.01 g of this dispersion was mixed with a solution of 13.5 g of Fluorad™ Fluorochemical Acrylate FX-13 (available from 3M Co., St. Paul, MN) and 1.5 g of methacrylic acid in acetone (65 ml) and the mixture heated to 65°C. The reaction mixture was deoxygenated by a N₂ purge for approximately 5 minutes and tert-butyl hydroperoxide (0.215 g of 70% active compound) added and heating, with stirring, continued for 22 hours. The resulting reaction mixture was free from monomer odor and had a 3:1 ratio of grafted polymer/silica at a 20% solids suspension.

A wooden tongue depressor was dipped into the suspension, excess liquid removed by wiping the depressor with a paper tissue, and the depressor allowed to dry at least 15 minutes at room temperature before testing for repellency. Oil repellency on the treated depressor demonstrated beading through oil # 3 and hold out through oil # 6. Aqueous kit repellency for the treated depressor demonstrated beading through oil # 1 and hold out through oil # 3. The depressor demonstrated excellent dynamic repellency in a stream of water impinging on the depressor from a height of approximately 30 cm.

Example 43

A treatment solution was prepared substantially according to the procedure of Example 41 wherein an aqueous solution of FCSi (1.0 g FCSi in 49 g D.I. water) was added to the polymer grafted silica of Example 1, previously diluted with D.I. water to 2% solids (50 g), to produce 100 g of an approximately 2%

Tongue depressors were treated with the redispersed FCSi/polymer grafted silica samples and tested for oil kit repellency and dynamic water repellency essentially as described in Example 41. Depressors treated with the redispersions demonstrated oil kit repellency bead up through 4 and holdout through 6 and good dynamic water repellency.

Acetone (65.6g) was added to a 17% solids dispersion of Precursor P-1 (29.4 g) to obtain an approximately 5% solids, mixed solvent colloidal dispersion. Methyl methacrylate (5.0g) was added dropwise to the dispersion while stirring, the resulting mixture heated to 65°C under a N₂ atmosphere and then deoxygenated by a N₂ purge for 5 minutes. t-Butyl hydroperoxide (.071g of 70% solids aqueous solution, available from Aldrich Chemical Co.) was added and the reaction mixture heated at 65°C, under a N₂ atmosphere, for 24 hours to produce a clear suspension of polymer grafted silica. A solution of FCSi (0.250g) in acetone (4.75g) was added drop wise to an aliquot (5g) of the poly(methylmethacrylate) grafted

silica and the resulting mixture heated, with stirring, at 65°C for 24 hours.

5 A tongue depressor was treated with the resulting FCSi/polymer graft suspension and tested for oil kit and dynamic water repellency essentially as described in Example 41. The treated depressor demonstrated oil bead up through 4 and holdout through 6 and excellent dynamic water repellency.

10

Example 45

D.I. water (50 g) was added to a SnO₂ suspension (50 g, 88SN-123, 20% solids, <10 nm particle size, pH 10, available from Nalco Chemical Co.) to produce an approximately 10% solids colloidal suspension.

15 Mercaptopropyl trimethoxy silane (0.25g, 0.13 mM/g) was added to this suspension and the mixture heated, with stirring, at 80°C for 16h.

20

Hydroxy ethyl acrylate (2.0 g, HEA, available from Aldrich Chemical Co.) was added to a 20 g aliquot of the resulting sol, the mixture heated to 60°C, degassed with N₂, and t-butylhydroperoxide (30 mg, 70% solids aqueous solution) added to the reaction mixture. The mixture was heated at 65°C for 16h to give a slightly opaque, stable suspension of SnO₂-g-HEA (1:1) graft.

25

Various modifications and alterations of this invention will become apparent to those skilled in the art without departing from the scope and spirit of the invention.

Claims

1. A fluorochemical treatment composition comprising (a) a sol comprising a colloidal dispersion of substantially discrete, substantially monodisperse in size, polymer-grafted, inorganic microparticles in liquid; and (b) at least one fluorochemical.

2. The composition of Claim 1 wherein said liquid is selected from the group consisting of water, at least one organic liquid, and aqueous solutions of at least one organic liquid.

3. The composition of Claim 1 wherein said inorganic microparticles comprise metal oxide selected from the group consisting of silica, titania, alumina, zirconia, vanadia, chromia, iron oxide, antimony oxide, tin oxide, and mixtures thereof.

4. The composition of Claim 1 wherein said inorganic microparticles have an average particle diameter in the range of from about 1 nanometer to about 200 nanometers.

5. The composition of Claim 1 wherein said inorganic microparticles are grafted with polymer comprising polymerized units of at least one free-radically polymerizable, ethylenically-monounsaturated monomer.

6. The composition of Claim 1 wherein said inorganic microparticles are grafted with polymer by means of at least one coupling agent which comprises at least one functional group which can function as a chain transfer site for free radical polymerization.

7. The composition of Claim 1 wherein said fluorochemical comprises at least one fluoroaliphatic or fluoroalicyclic group.

5 8. The composition of Claim 1 wherein said fluorochemical is capable of chemical reaction or strong physical interaction with chemical groups on the surface of said inorganic microparticles.

10 9. The composition of Claim 1 in substantially liquid-free form.

10. A treatment composition comprising a sol comprising a colloidal dispersion of substantially
15 discrete, substantially monodisperse in size, polymer-grafted, inorganic microparticles in liquid.

11. A treatment composition comprising a sol comprising a colloidal dispersion of substantially
20 discrete, substantially monodisperse in size, polymer-grafted, silica microparticles in water, wherein said inorganic microparticles are grafted with polymer comprising polymerized units of at least one monomer selected from the group consisting of methacrylic acid
25 and fluorochemical-containing acrylates.

12. A process for imparting durable water and oil repellency, stain release, and dry soil resistance properties to a substrate comprising the step of
30 contacting a substrate with the composition of Claim 1.

13. A process for imparting durable water and oil repellency, stain release, and dry soil resistance properties to a substrate comprising the steps of (a)
35 contacting a substrate with a sol comprising a

colloidal dispersion of substantially discrete, substantially monodisperse in size, polymer-grafted, inorganic microparticles in liquid; and (b) contacting said substrate with at least one fluorochemical.

5

14. A process for imparting durable stain release and/or dry soil resistance properties to a substrate comprising the step of contacting a substrate with the composition of Claim 10.

10

15. A treated article comprising the composition of Claim 1 or the corresponding substantially liquid-free composition.

15

16. A treated article comprising the composition of Claim 10 or the corresponding substantially liquid-free composition.

INTERNATIONAL SEARCH REPORT

International Application No
PC1/US 96/08302

A. CLASSIFICATION OF SUBJECT MATTER
IPC 6 D06M11/79 D06M15/277 D06M15/657

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)
IPC 6 D06M

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	EP,A,0 121 078 (BAYER AG) 10 October 1984 see the whole document ---	1-16
A	EP,A,0 550 915 (IDEMITSU KOSAN CO) 14 July 1993 see page 3, line 10 - line 27 see page 4, line 1 - line 18; claims ---	1-16
A	US,A,4 617 057 (PLUEDDEMANN EDWIN P) 14 October 1986 see column 5, line 26 - line 46; claims ---	1-16
A	WO,A,93 01348 (MINNESOTA MINING & MFG) 21 January 1993 see page 12, line 7 - line 26; claims --- -/--	1-16

☒ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

* Special categories of cited documents :

- *A* document defining the general state of the art which is not considered to be of particular relevance
- *E* earlier document but published on or after the international filing date
- *L* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- *O* document referring to an oral disclosure, use, exhibition or other means
- *P* document published prior to the international filing date but later than the priority date claimed

T later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

X document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

Y document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.

G document member of the same patent family

Date of the actual completion of the international search

25 November 1996

Date of mailing of the international search report

29. 11. 96

Name and mailing address of the ISA
European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+ 31-70) 340-2040, Tx. 31 651 epo nl,
Fax: (+ 31-70) 340-3016

Authorized officer

Blas, V

INTERNATIONAL SEARCH REPORT

International Application No

PC1/US 96/08302

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	EP,A,0 457 939 (BASF CORP) 27 November 1991 see page 3, line 3 - line 18; claims ---	1,4-6,10
A	GB,A,1 159 124 (JOSEPH GEOGE SPITZER ET AL.) 23 July 1969 see page 2, line 92 - page 3, line 71; claims ---	1-5, 10-16
A	EP,A,0 433 906 (TOSHIBA SILICONE ;JAPAN SYNTHETIC RUBBER CO LTD (JP)) 26 June 1991 see the whole document ---	1,3,5,6, 11
A	DATABASE WPI Section Ch, Week 8741 Derwent Publications Ltd., London, GB; Class A60, AN 87-290933 XP002019405 & SU,A,1 298 213 (IVANCHEV S S) , 23 March 1987 see abstract -----	1,3,5,6, 11

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PC1/US 96/08302

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP-A-0121078	10-10-84	DE-A- 3307420 US-A- 4781844	13-09-84 01-11-88
EP-A-0550915	14-07-93	JP-A- 5279657 JP-A- 5295354 US-A- 5431852	26-10-93 09-11-93 11-07-95
US-A-4617057	14-10-86	CA-A- 1285678 DE-A- 3680595 EP-A- 0207282 JP-A- 61278574 KR-B- 9404199	02-07-91 05-09-91 07-01-87 09-12-86 17-05-94
WO-A-9301348	21-01-93	AU-B- 653629 AU-A- 2299292 DE-D- 69210608 EP-A- 0593617 JP-T- 6509146 US-A- 5370919	06-10-94 11-02-93 13-06-96 27-04-94 13-10-94 06-12-94
EP-A-0457939	27-11-91	AT-T- 109188 DE-D- 69011095 DE-T- 69011095	15-08-94 01-09-94 23-02-95
GB-A-1159124	23-07-69	NL-A- 6613944 FR-A- 1494683 BE-A- 687564 DE-A- 1619158 LU-A- 52071	04-04-68 14-12-67 01-03-67 11-03-71 29-11-66
EP-A-0433906	26-06-91	DE-D- 69021181 DE-T- 69021181 JP-A- 4270710 KR-B- 9400018 US-A- 5218014	31-08-95 25-01-96 28-09-92 05-01-94 08-06-93